ELSEVIER

Contents lists available at ScienceDirect

# **Bioorganic & Medicinal Chemistry**

journal homepage: www.elsevier.com/locate/bmc



# Nakijiquinones E and F, new dimeric sesquiterpenoid quinones from marine sponge

Yohei Takahashi, Takaaki Kubota, Jun'ichi Kobayashi \*

Graduate School of Pharmaceutical Sciences, Hokkaido University, Sapporo 060-0812, Japan

#### ARTICLE INFO

Article history: Received 2 April 2008 Revised 13 June 2008 Accepted 31 October 2008 Available online 5 November 2008

Keywords: Marine sponge Dimeric sesquiterpenoid quinones Nakijiquinones E and F

#### ABSTRACT

Two new dimeric sesquiterpenoid quinones, nakijiquinones E(1) and F(2), have been isolated from an Okinawan marine sponge, and the structures and relative stereochemistry of  $\mathbf{1}$  and  $\mathbf{2}$  were elucidated on the basis of the spectral data. Nakijiquinones E(1) and F(2) were the first dimeric sesquiterpenoid quinones possessing a 3-aminobenzoate moiety.

© 2008 Elsevier Ltd. All rights reserved.

# 1. Introduction

Marine sponges contain a number of unique secondary metabolites with a diversity of biological activities. During our search for bioactive metabolites from marine organisms, we recently isolated new sesquiterpenoid quinones, metachromins L–T, from an Okinawan sponge *Spongia* sp. (SS-1037). Further investigation of extracts of another lot of the sponge resulted in the isolation of two new dimeric sesquiterpenoid quinones, nakijiquinones E (1) and F (2). Here, we describe the isolation and structure elucidation of 1 and 2.

#### 2. Results and discussion

The sponge collected off Unten Port, Okinawa, was extracted with MeOH. The extracts were partitioned between EtOAc and water. EtOAc-soluble materials were purified by silica gel  $C_{18}$  column chromatographies followed by  $C_{18}$  HPLC (Wakosil-II 5C18 AR, CH<sub>3</sub>CN/H<sub>2</sub>O/TFA) to afford nakijiquinones E (**1**, 0.0025%, wet weight) and F (**2**, 0.0009%) together with known related sesquiterpenoids, dictyoceratins A-C,<sup>5,6</sup> isospongiaquinone,<sup>7</sup> 6'-hydroxy-4'-methoxyavarone,<sup>8</sup> neoavarol,<sup>9</sup> nakijiquinones A-D,<sup>10,11</sup> and *endo*-olefin isomer at C-3 of smenospongine.<sup>12</sup>

O CO<sub>2</sub>Me
HO 18 HO 18' 15'
15 16 20 N 18' 16' 15'
H HO 14' H 10' 15'
11' 10' 5'
12' 11'

Nakijiquinone E (1) was obtained as a red amorphous solid and the molecular formula was established to be  $C_{44}H_{59}NO_7$  by HRESIMS data [m/z 736.42002 (M+Na)<sup>+</sup>,  $\varDelta$  +1.1 mmu]. IR absorptions implied the presence of OH and/or NH (3280 cm<sup>-1</sup>), carboxy (1670 cm<sup>-1</sup>), and conjugated carbonyl (1640 and 1590 cm<sup>-1</sup>) functionalities. UV

2

<sup>\*</sup> Corresponding author. Tel.: +81 11 706 3239; fax: +81 11 706 4989. E-mail address: jkobay@pharm.hokudai.ac.jp (J. Kobayashi).

absorptions (313 and 494 nm) suggested the presence of quinone chromophore. The HRESIMS data and the  $^1\mathrm{H}$  and  $^{13}\mathrm{C}$  NMR spectra suggested that 1 was a dimeric sesquiterpenoid quinone.

The  $^1\text{H}$  NMR spectrum of  $\mathbf{1}$  in CDCl $_3$  showed signals due to a secondary methyl ( $\delta_{\text{H}}$  0.99), three tertiary methyls ( $\delta_{\text{H}}$  1.55, 1.02, and 0.85), and a singlet olefinic proton ( $\delta_{\text{H}}$  5.15). These data suggested the presence of a tetramethyl decaline moiety with an endo olefin (C-1–C-14) in  $\mathbf{1}$ , while signals due to a secondary methyl ( $\delta_{\text{H}}$  0.99), an exomethylene ( $\delta_{\text{H}}$  4.46 and 4.41), and two tertiary methyls ( $\delta_{\text{H}}$  1.07 and 0.85) implied the presence of a trimethyl decaline moiety with an exomethylene (C-1′–C-14′) in  $\mathbf{1}$ . Inspection of the  $^1\text{H}$ - $^1\text{H}$  COSY and HMBC spectra of  $\mathbf{1}$  revealed the presence of these two decaline moieties (C-1–C-14 and C-1′–C-14′) as shown in Figure 1.

The presence of a 2-amino-5-hydroxy-benzoquinone moiety was deduced from the chemical shifts<sup>10</sup> of C-16–C-21 ( $\delta_C$  183.0, 180.0. 156.0. 148.9. 115.1. and 97.0) in 1. This was confirmed by HMBC correlations of H-19 to C-17 and C-21, H<sub>2</sub>-15 to C-16, C-17 and C-21, and 20-NH to C-19 and C-21. The 2-amino-5-hydroxybenzoquinone moiety (C-16-C-23) and a decaline (C-l-C-14) were suggested to be connected between C-9 and C-16 through C-15 on the basis of HMBC correlations of H<sub>2</sub>-15 to C-9 and C-10, while the chemical shifts of C-16'-C-23' ( $\delta_{\rm C}$  170.1, 156.6, 156.1, 133.2, 117.5, 110.6, 105.3, and 52.3) and HMBC correlations for  $H_2$ -15' to C-16', C-17', and C-21', H-21' to C-17', C-19', C-21', and C-22', 17'-OH to C-17', 20-NH to C-18', 20-NH to C-18', and H<sub>3</sub>-23' to C-22' revealed the presence of methyl 3-amino-2,4-dihydroxybenzoate moiety (C-16'-C-22' and C-23') in **1** as shown in Figure 1. The connection between C-9' and C-16' through C-15' was implied by HMBC crosspeaks for H<sub>2</sub>-15 to C-9' and C-10'. Thus, the gross structure of nakiiiquinone E was elucidated to be 1.

The relative stereochemistry of the two decaline moieties in nakijiquinone E (1) were elucidated on the basis of NOESY correlations as shown in Figures 2 and 3. The a-configuration of H-10 and  $\beta$ -configurations of C-12, C-13, and C-14 were deduced from NOESY correlations of H-8/H-10, H-10/H<sub>2</sub>-15, and H<sub>3</sub>-12/H<sub>3</sub>-14, while NOESY correlations of H-8'/H-10', H-10'/H<sub>2</sub>-15', and H<sub>3</sub>-12'/H<sub>3</sub>-14' revealed that three methyl groups (Me-12', Me-13', and Me-14') were all  $\beta$ -oriented and H-10 was  $\alpha$ -oriented.

Nakijiquinone F (2) was obtained as a red amorphous solid and the molecular formula was established to be  $C_{44}H_{59}NO_7$  by HRE-SIMS data [m/z 736.41828 (M+Na)<sup>+</sup>,  $\Delta$  –0.6 mmu]. IR and UV spectra of 2 were almost the same as those of 1, suggesting that nakijiquinone F (2) was an analog of 1.  $^{1}H$  and  $^{13}C$  NMR spectra of 2 differed from those of 1 in lacking of signals for an exomethylene present in 1. The  $^{1}H$  NMR spectrum of 2 showed signals for olefinic protons ( $\delta_H$  5.15, 2H) and olefinic methyls ( $\delta_H$  1.55, and 1.53, 3H each). These data suggested that nakijiquinone F (2) possessed two tetramethyl decaline rings with an endo olefin. Analysis

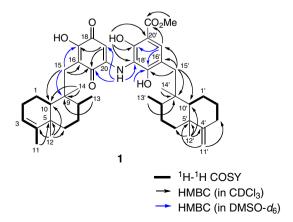
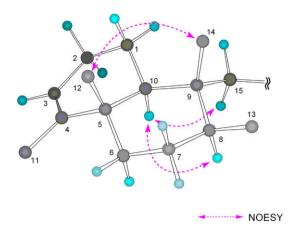
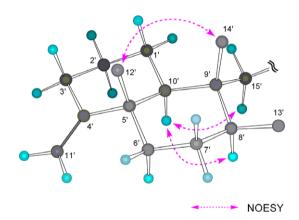


Figure 1. Selected 2D NMR correlations for nakijiquinone E (1).



**Figure 2.** Selected NOESY correlations and relative stereochemistry for C-1–C-15 moiety of nakijiquinone E (1) (hydrogen atoms of methyl groups were omitted).



**Figure 3.** Selected NOESY correlations and relative stereochemistry for C-1′–C-15′ moiety of nakijiquinone E (1) (hydrogen atoms of methyl groups were omitted).

of  ${}^{1}\text{H} - {}^{1}\text{H}$  COSY and HMBC spectra of nakijiquinone F indicated the gross structure to be **2** (Fig. 4). The relative stereochemistries of the two decaline rings in **2** were elucidated to be the same as those of C-1–C-15 moiety of **1** on the basis of the NOESY data (Fig. 5).

Nakijiquinones E (1) and F (2) were the first dimeric sesquiterpenoid quinones possessing a 3-aminobenzoate moiety, though some dimeric sesquiterpenoid quinones from the sponges *Dysidea* sp.  $^{13-15}$  have been reported so far. Nakijiquinones E (1) and F (2) did not show cytotoxicity against murine leukemia P388 and L1210, and KB human epidermoid carcinoma cells ( $IC_{50} > 10 \mu g/mL$ ).

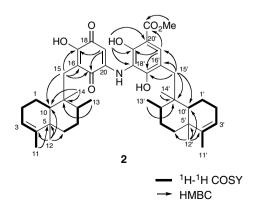
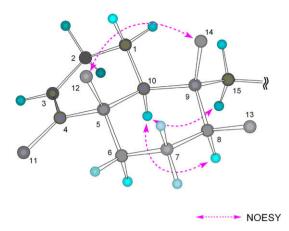


Figure 4. Selected 2D NMR correlations for nakijiquinone F (2).



**Figure 5.** Selected NOESY correlations and relative stereochemistry for C-1–C-15 moiety of nakijiquinone F (2) (hydrogen atoms of methyl groups were omitted).

#### 3. Experimental

#### 3.1. General

Optical rotation was recorded on a JASCO P-1030 polarimeter. IR and UV spectra were recorded on JASCO FT/IR-230 and Shimadzu UV-1600PC spectrophotometer, respectively.  $^1\mathrm{H}$  and  $^{13}\mathrm{C}$ 

NMR spectra were recorded on a Bruker AMX-600 spectrometers using 2.5 mm micro cells (Shigemi Co., Ltd) for CDCl<sub>3</sub> and DMSO- $d_6$ , respectively. The 7.26 and 77.0 ppm resonances of residual CHCl<sub>3</sub> and the 2.49 and 49.8 ppm resonances of residual DMSO were used as internal references for  $^1$ H and  $^{13}$ C NMR spectra, respectively. ESI mass spectra were obtained on a JEOL JMS-700TZ spectrometer. Molecular mechanics calculations were performed by using CS Chem 3D Ultra ver. 7.0 (MM2 force field).  $^{16}$ 

#### 3.2. Extraction and isolation

The sponge (0.6 kg, wet weight) was extracted with MeOH, and the extract (20.2 g) was partitioned between EtOAc and H<sub>2</sub>O. EtOAc-soluble materials (2.2 g) was purified by a silica gel column (hexane/acetone), a C<sub>18</sub> column (MeOH/H<sub>2</sub>O) and silicagel columns (hexane/CHCl<sub>3</sub>, and then hexane/EtOAc) followed by C<sub>18</sub> HPLC (Wakosil-II 5C18 AR, Wako Pure Chemical Ind., Ltd,  $10 \times 250$  mm; eluent CH<sub>3</sub>CN/H<sub>2</sub>O/TFA, 95:5:0.05; flow rate, 2.0 mL/min; UV detection at 300 nm) to afford nakijiquinones E (1) (1, 2.0 mg,  $t_R$  36 min) and F (2, 0.7 mg,  $t_R$  38 min).

### 3.3. Nakijiquinone E (1)

Red amorphous solid;  $[\alpha]_{\rm D}^{23}$  +54 (c 0.25, CHCl<sub>3</sub>); IR (film)  $v_{\rm max}$  3280, 1670, 1640, 1590, 1510, 1440, 1380, 1340, and 1210 cm<sup>-1</sup>;

**Table 1**<sup>1</sup>H and <sup>13</sup>C NMR Data of Nakijiquinone E (1) in CDCl<sub>3</sub>

Position	$\delta_{c}$	$\delta_{\rm H}$ (m, J in Hz)	HMBC	Position	$\delta_{c}$	$\delta_{ ext{H (m, J in Hz)}}$	HMBC
1	20.0 CH <sub>2</sub>	2.06 (m) 1.49 (m)		1′	23.1 CH <sub>2</sub>	2.01 (m) 1.59 (m)	
2	27.1 CH <sub>2</sub>	2.04 (m) 1.95 (m)		2′	27.9° CH <sub>2</sub>	1.90 (m) 1.2–1.3 (m)	
3	120.9 CH	5.15 (br s)	2, <sup>h</sup> 5, <sup>h</sup> 11 <sup>h</sup>	3′	33.0 CH <sub>2</sub>	2.35 (ddd, 13.5, 13.5, 4.8)	
4	144.2 C					2.11 (m)	
5	38.5 C			4′	159.7 C		
6	35.9 CH <sub>2</sub>	1.64 (m) 1.11 (m)		5′ 6′	40.2 C 36.5 <sup>f</sup> CH <sub>2</sub>	1.50 (m)	
7	28.0 CH <sub>2</sub>	1.38 (m) 1.34 (m)		7′	27.6 CH <sub>2</sub>	1.21 (m) 1.42 <sup>a</sup> (m)	
8	37.9 CH	1.32 (m)		8′	36.5 <sup>f</sup> CH	1.25 (m)	
9	42.8 C			9′	41.9 C		
10	47.7 CH	1.11 (d, 12.0)		10′	48.1 CH	0.91 (d, 11.8)	
11	18.2 CH₃	1.55 <sup>b</sup> (br s)	3, 4, 5	11'	103.1 CH <sub>2</sub>	4.46 (s)	3', 5'
12	20.2 CH <sub>3</sub>	1.02 <sup>b</sup> (s)	4, 5, 6, 10			4.41 (s)	
13	17.5 <sup>d</sup> CH <sub>3</sub>	0.99 <sup>b</sup> (d, 6.3)	7, 8, 9	12′	20.5 CH <sub>3</sub>	1.07 <sup>b</sup> (s)	4', 5', 6', 10'
14	17.3 <sup>e</sup> CH₃	0.85 <sup>b</sup> (s)	8, 9, 10, 15	13′	17.7 <sup>d</sup> CH₃	0.99 <sup>b</sup> (d, 6.3)	7', 8', 9'
15	32.6 CH <sub>2</sub>	2.61 (d, 13.9) 2.47 (d, 13.9)	8, <sup>h</sup> 9, <sup>h</sup> 10, 16, 17, <sup>h</sup> 21 8, <sup>h</sup> 9, <sup>h</sup> 10, 16, 17, <sup>h</sup> 21	14′ 15′	17.4 <sup>e</sup> CH₃ 36.6 <sup>f</sup> CH₂	0.85 <sup>b</sup> (s) 2.59 <sup>a</sup> (br s)	8', 9', 10', 15' 8', <sup>h</sup> 9', 10', 14', 16', 17', 21'
16	115.1 C			16′	117.5 C		
17	156.0 C			17′	156.6 C		
18	180.0 C			18′	110.6 <sup>g</sup> C		
19	97.0 CH	5.21 (s)	17, 21	19′	156.1 C		
20	148.9 C			20′	105.3 <sup>g</sup> C		
21	183.0 C			21′	133.2 CH	7.54 (s)	15′, 17′, h 19′, 22′
20-NH		5.8–6.4 (br s)	19, <sup>h</sup> 21, <sup>h</sup> 18 <sup>,h</sup>	22′ 23′ 17′-OH 19′-OH	170.1 C 52.3 CH <sub>3</sub>	3.91 <sup>b</sup> (s) 8.45 11.06 (s)	22' 16' <sup>h</sup> 18', 19', 20'

<sup>&</sup>lt;sup>a</sup> 2H.

<sup>&</sup>lt;sup>b</sup> 3Н.

 $<sup>^{</sup>c-g}$  Interchangeable.

h In DMSO-d<sub>6</sub>.

**Table 2**  $^{1}$ H and  $^{13}$ C NMR Data of Nakijiquinone F (**2**) in CDCl<sub>3</sub>

Position	$\delta_{C}$	$\delta_{\rm H}$ (m, J in Hz)	НМВС	Position	$\delta_{C}$	$\delta_{\rm H}$ (m, J in Hz)	НМВС
1	20.0° CH <sub>2</sub>	2.08 (m) 1.49 (m)		1′	19.7 CH <sub>2</sub>	1.99 (m) 1.64 (m)	
2	27.1 CH <sub>2</sub>	2.04 (m) 1.92 (m)		2′ 3′	26.0 CH <sub>2</sub> 120.4 CH	2.11 <sup>a</sup> (m) 5.15 (br s)	
3	120.9 CH	5.15 (br s)		4′	144.4 C		
4	144.2 C			5′	38.3 C		
5	38.5 C			6′	35.9 <sup>d</sup> CH <sub>2</sub>	1.60 (m)	
6	35.9 <sup>d</sup> CH <sub>2</sub>	1.65 (m) 1.11 (m)		7′	27.7 CH <sub>2</sub>	0.94 (m) 1.37 <sup>a</sup> (m)	
7	28.0 CH <sub>2</sub>	1.37 <sup>a</sup> (m)		8′	36.2 CH	1.22 (m)	
8	37.9 CH	1.32 (m)		9′	41.5 C		
9	42.8 C			10′	45.6 CH	1.17 (d, 11.8)	
10	47.7 CH	1.11 (d, 11.6)		11′	18.1 CH₃	1.53 <sup>b</sup> (s)	3', 4', 5'
11	18.2 CH₃	1.55 <sup>b</sup> (br s)	3, 4, 5	12′	20.0 <sup>c</sup> CH₃	1.03 <sup>b,g</sup> (s)	4', 5', 6', 10'
12	20.2 CH <sub>3</sub>	1.02 <sup>b,g</sup> (s)	4, 5, 6, 10	13′	17.6e CH <sub>3</sub>	1.00 <sup>b</sup> (d, 6.3)	7', 8', 9'
13	17.7e CH₃	1.00 <sup>b</sup> (d, 5.7)	7, 8, 9	14′	17.3 <sup>f</sup> CH₃	0.86 <sup>b</sup> (s)	8', 9', 10', 15'
14	17.5 <sup>f</sup> CH₃	0.86 <sup>b</sup> (s)	8, 9, 10, 15	15′	36.7 CH <sub>2</sub>	2.70 (d, 14.5)	9', 10', 16', 17', 21'
15	32.6 CH <sub>2</sub>	2.62 (d, 13.7) 2.49 (d, 13.7)	8, 9, 10, 16, 17, 21 9, 10, 16, 17, 21	16′	115.1 C	2.61 (d, 14.5)	8', 9'
16	115.1 C	( , , , , ,	, , , , ,	17′	156.6 C		
17	155.9 C			18′	105.4 C		
18	180.0 C			19′	156.2 C		
19	97.1 CH	5.24 (s)	17, 21	20′	105.4 C		
20	148.9 C			21′	133.4 CH	7.60 (s)	15', 17', 19', 22'
21	183.0 C			22′	170.2 C		
20-NH		5.88 (br)		23′ 19′-OH	52.2 CH₃	3.92 <sup>b</sup> (s) 11.07 (s)	22' 18', 19', 20'

а 2H.

UV (MeOH)  $\lambda_{\rm max}$  265 ( $\varepsilon$  14,300, sh), 313 (17,500), and 494 nm (1700);  $^{1}$ H and  $^{13}$ C NMR data see Table 1; ESIMS m/z 736 (M+Na) $^{+}$  and 1449.8 (2M+Na) $^{+}$ ; HRESIMS m/z 736.42002 [(M+Na) $^{+}$ , calcd for C<sub>44</sub>H<sub>59</sub>NO<sub>7</sub>Na, 736.41892, error +1.49 ppm].

## 3.4. Nakijiquinone F (2)

Red amorphous solid;  $[\alpha]_D^{25}$  +44 (c 0.2, CHCl3); IR (film)  $v_{\rm max}$  3270, 1670, 1640, 1590, 1500, 1440, 1380, 1340, and 1210 cm<sup>-1</sup>; UV (MeOH)  $\lambda_{\rm max}$  265 ( $\epsilon$  17,300, sh), 313 (21,500), and 490 nm (2200);  $^1$ H and  $^{13}$ C NMR data see Table 2; ESIMS (positive) m/z 714 (M+H) $^+$  and 736 (M+Na) $^+$ ; HRESIMS m/z 736.41828 [(M+Na) $^+$ , calcd for  $C_{44}H_{59}NO_7$ , 736.41892, error -0.87 ppm].

# Acknowledgments

We thank Ms. S. Oka, Center for Instrumental Analysis, Hokkaido University, for measurements of ESIMS. This work was partly supported by a grant from the Uehara Memorial Foundation and a Grant-in-Aid for Scientific Research from the Ministry of Education, Culture, Sports, Science and Technology of Japan.

#### References and notes

- Blunt, J. W.; Copp, B. R.; Hu, W.-P.; Munro, M. H. G.; Northcote, P. T.; Prinsep, M. R. Nat. Prod. Rep. 2008, 25, 35.
- Nishi, T.; Kubota, T.; Fromont, J.; Sasaki, T.; Kobayashi, J. Tetrahedron 2008, 64, 3127.
- 3. Takahashi, Y.; Kubota, T.; Fromont, J.; Kobayashi, J. *Tetrahedron* **2007**, 63, 8770.
- Takahashi, Y.; Yamada, M.; Kubota, T.; Fromont, J.; Kobayashi, J. Chem. Pharm. Bull. 2007, 55, 1731.
- 5. Nakamura, H.; Deng, S.; Kobayashi, J.; Ohizumi, Y.; Hirata, Y. *Tetrahedron* **1986**, 42, 4197.
- Kushlan, D. M.; Faulkner, D. J.; Parkanyi, L.; Clardy, J. Tetrahedron 1989, 45, 3307.
- Kazlauskas, R.; Murphy, P. T.; Warren, R. G.; Wells, R. J.; Blount, J. F. Aust. J. Chem. 1978, 31, 2685.
- 8. Loya, S.; Hizi, A. FEBS Lett. 1990, 269, 131-134.
- 9. Iguchi, K.; Sahashi, A.; Kohno, J.; Yamada, Y. Chem. Pharm. Bull. 1990, 38, 1121.
- Shigemori, H.; Madono, T.; Sasaki, T.; Mikami, Y.; Kobayashi, J. *Tetrahedron* 1994, 50, 8347.
- 11. Kobayashi, J.; Madono, T.; Shigemori, H. Tetrahedron 1995, 51, 10867.
- Utkina, N.; Denisenko, V. A.; Scholokova, O. V.; Makarchenko, A. E. J. Nat. Prod. 2003, 66, 1263–1265.
- Rodriguez, A. D.; Yoshida, W. Y.; Scheuer, P. J. *Tetrahedron* **1990**, *46*, 8025.
- Alvi, K. A.; Diaz, M. C.; Crews, P.; Slate, D. L.; Lee, R. H.; Moretti, R. J. Org. Chem. 1992, 57, 6604.
- 15. Carney, J. R.; Scheuer, J. *Tetrahedron Lett.* **1993**, 23, 3727.
- 16. Allinger, N. L. J. Am. Chem. Soc. 1977, 99, 8127.

<sup>&</sup>lt;sup>ь</sup> 3Н.

c-g Interchangeable.